

CENTRAL BOARD OF SECONDARY EDUCATION

2, Community Centre, Preet Vihar, Delhi-110092

Assessment of Practical Skills in Science

Class X

(Effective from March, 2010 Examination)

CBSE, Delhi - 110092

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हम, भारत के लोग, भारत को एक (सम्पूर्ण प्रभुत्व—संपन्न समाजवादी पंथनिरपेक्ष लोकतंत्रात्मक गणराज्य) बनाने के लिए तथा उसके समस्त नागरिकों को :

> सामाजिक, आर्थिक और राजनैतिक न्यय, विचार, अभिव्यक्ति, विश्वास, धर्म और उपासना की स्वतंत्राता, प्रतिष्ठा और अवसर की समता प्राप्त कराने के लिए,

तथा उन सब में

व्यक्ति की गरिमा और² (राष्ट्र की एकता और अखण्डता) सुनिश्चित करने वाली बंधुता बढ़ाने के लिए दृढ़संकल्प होकर अपनी इस संविधान सभा में आज तारीख 26 नवम्बर,, 1949 ई० को एतद्द्वारा इस संविधान को अंगीकृत, अधिनियमित और आत्मार्पित करते हैं।

- 1. संविधान (बयालीसवां संशोधन) अधिनियम, 1976 की धारा 2 द्वारा (3.1.1977 से) "प्रभुत्व—संपन्न लोकतंत्रात्मक गणराज्य" के स्थान पर प्रतिस्थापित।
- 2. संविधान (बयालीसवां संशोधन) अधिनियम, 1976 की धारा 2 द्वारा (3.1.1977 से) "राष्ट्र की एकता" के स्थान पर प्रतिस्थापित।

Hkkx 4 d eny dùk0;

5-1 d- eny dùld); & भारत के प्रत्येक नागरिक का यह कर्त्तव्य होगा कि वह-

- (क) संविधान का पालन करे और उसके आदर्शों, संस्थाओं, राष्ट्र ध्वज और राष्ट्रगान आदर करें;
- (ख) स्वतंत्रता के लिए हमारे राष्ट्रीय आंदोलन को प्रेरित करने वाले उच्च आदर्शों को हृदय में संजोए रखें और उनका पालन करें;
- (ग) भारत की प्रभृता, एकता और अखंडता की रक्षा करें और उसे अक्ष्णण रखें;
- (घ) देश की रक्षा करें और आह्वान किए जाने पर राष्ट्र की सेवा करें;
- (ङ) भारत के सभी लोगों में समरसता और समान भ्रातृत्व की भावना का निर्माण करें जो धर्म, भाषा और प्रदेश या वर्ग पर आधारित सभी भेदभाव से परे हों, ऐसी प्रथाओं का त्याग करें जो स्त्रियों के सम्मान के विरुद्ध हैं:
- (च) हमारी सामाजिक संस्कृति की गौरवशाली परंपरा का महत्व समझें और उसका परिरक्षण करें;
- (छ) प्राकृतिक पर्यावरण की जिसके अंतर्गत वन, झील, नदी और वन्य जीव हैं, रक्षा करें और उसका संवर्धन करें तथा प्राणि मात्रा के प्रति दयाभाव रखें:
- (ज) वैज्ञानिक दृष्टिकोण, मानववाद और ज्ञानार्जन तथा सुधार की भावना का विकास करें;
- (झ) सार्वजनिक संपत्ति को सुरक्षित रखें और हिंसा से दूर रहें;
- (ञ) व्यक्तिगत और सामूहिक गतिविधियों के सभी क्षेत्रों में उत्कर्ष की ओर बढ़ने का सतत प्रयास करें जिससे राष्ट्र निरंतर बढ़ते हुए प्रयत्न और उपलिब्ध की नई ऊँचाईयों को छू लें।

THE CONSTITUTION OF INDIA

Preamble

We, The people of India, having solemnly resolved to constitute India into a¹ (Sovereign Socialist Secular democratic republic) and to secure to all its citizens:

Justice, Social, Economic and Politicial;

Liberty of thought, expression, belief, faith and worship;

Equiality of status and of opportunity; and to promote among them all

Fraternity assuring the dignity of the individual and the² (unity and integrity of the Nation);

In our Constitutent Assembly this twenty-sixth day of November, 1949, do hereby adopt, enact and give to ourselves this constitution.

- 1. Subs. by the Constitution (Forty-second Amendment) Act, 1976, sec. 2, for "Sovereign Democratic Republic (w.e.f. 3.1.1977)".
- 2. Subs. by the Constitution (Forty-second Amendment) Act, 1976, sec. 2, for "Unity of the Nation (w.e.f. 3.1.1977)".

THE CONSTITUTION OF INDIA

Chapter IV A

Fundamental Duties

Article 51-A

Fundamental Duties- It shall be the duty of every citizen of India-

- (a) to abide the Constitution and respect its ideals and institutions, the National Flag and the National Anthem;
- (b) to cherish and follow the noble ideals which inspired our national struggle for freedom;
- (c) to uphold and protect the sovereignty, unity and integrity of India;
- (d) to defend the countlry and render national service when called upon to do so;
- to promote harmony and the spirit of common brotherhood amongst all the people of India transcending religious, linguistic and regional or sectional diversities; to renounce practices derogatory to the dignity of women;
- (f) to value and preserve the rich heritage of our composite culture;
- (g) to protect and improve the rich natural environment including forests, lakes, rivers, wild life and to have compassion for living creatures;
- (h) to develop the specific temper, humanism and the spirit of inquiry and reform;
- (i) to safeguard public property and to abjure violence;
- (j) to strive towards excellence in all spheres of individual and collective activity so that the national constantly rises to higher levels of endeavour and achievement.

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FOREWORD

The central role of experiments and practical work in school science curriculum is universally accepted. A balanced science curriculum should not only give due emphasis to both theory and experiments, but also integrate these complimentary aspects of the subject in teaching-learning process. Present-day science, as we all know, is the result of creative interplay of observations, experimentation and theoretical inferences.

The importance of practical work in science education has also been recognized and greatly emphasized in national policy of education for the past several decades. However, despite several laudable efforts in the past, experiments, by and large, have continued to be marginalized in the schools. The challenge arises due to combination of several factors. The Board has been keen to find out ways to promote laboratory work in the subject and introduce greater uniformity, objectivity and reliability in the assessment of practical work. With the objective of strengthening experimentation and skills through a written test also as part of its evaluation procedure in class IX during 2005-2006. The scheme was extended to class X from the academic session 2006-07 onwards. It is hoped that this initiative has given experiments their due place in subject and has promoted, an experimental culture in our school system. The present document has been revised for 2010 board examination and includes questions based on experiments included in class IX as well as class X practical syllabus.

I shall put on record my heartfelt and sincere thanks to Homi Bhabha Centre for Science Education, Bombay for extending its full academic expertise and providing all necessary facilities for development of this material during two six-day workshops organized by the Board at the centre. The immensely rich and creative contribution made to the development and finalization of this material by Prof. Arvind Kumar, Director, HBCSE is gratefully acknowledged. The honest efforts put in by Prof. Chitra Natrajan, Prof. S.C. Agarkar, Mr. V.G. Gambhir and other faculty members of the centre and thankfully acknowledged. Sincere thanks are also due to other eminent subject experts from IGNOU, Delhi University, School Systems in Delhi and Chennai who worked hard and made rich contributions to the development and subsequent revision of this material. My deep appreciation and thanks are also due to Mrs. C. Gurumurthy, Director (Academic), CBSE and Shri R.P. Sharma, Consultant, CBSE, Dr. Srijata Das, Education Officer (Science & Maths), for putting in their best efforts in development and subsequent revision and publication of this document.

It is hoped that concerted efforts will be put in by all schools and subject teachers to make best use of this document and make learning of science a joyful and meaningful experience for learners.

Feedback and suggestions from readers for further improvement of this document will be highly appreciated.

> VINEET JOSHI, I.A.S. CHAIRMAN

Introduction

The crucial role of experiments in school science curriculum is universally accepted. A good science curriculum must not only give balanced emphasis to both theory and experiments but also integrate these two essential and complementary aspects of science in the teaching-learning process. Modern science, as we all know, is the result of a creative interplay of experiments, observations and theoretical inference.

There are several ways in which experiments facilitate and improve the learning of science. First and foremost, experiments help students develop the right perspective of science, namely that science is not just a theoretical abstraction – it is an attempt to describe the working of the real world around us. A hypothesis or idea in science is acceptable only if observations and experiments confirm it. Second, experiments are among the most effective ways to generate interest in science. For many students, an apparently 'dry', 'uninteresting' fact of a theory textbook can become live and exciting when translated into an experiment. Third, experiments promote the basic skills and competencies of doing science: procedural and manipulative skills, observation skills, skills of representing and interpreting data and the accompanying conceptual and critical abilities. For these various reasons, promoting activity and experiment based learning has been at the heart of many efforts aimed at improving science education in our country.

Despite several laudable efforts in the past, experiments, by and large, have continued to be marginalized in our schools. There seem to be two principal difficulties. Firstly, experiments require a certain minimum infrastructure – a laboratory with some basic equipments and consumables on a recurring basis. Secondly, assessment of practical skills in science in a sound and objective manner is by no means an easy task. The difficulty multiplies manifold if assessment is to be carried out on a large scale. Thus lack of infrastructure and, more important, lack of reliable assessment have resulted in the unfortunate neglect of experimental work in most of the schools in India.

The Board's Initiative

The problem of neglect of experiments in our schools and of proper assessment of students in practical skills has always been a matter of great concern for the Central Board of Secondary Education. The problem assumes even greater importance for Class X, which is the terminal stage of secondary education. The Board has been keen to find out ways to promote laboratory work in our school system and has taken a number of initiatives in this direction. The Board tries to ensure that its affiliated schools have the necessary infrastructure to carry out experiments prescribed in the syllabus to Class IX and X. To make this feasible for all its schools, care is taken that the laboratory curriculum does not demand prohibitively costly equipment or other unrealistic requirements. As another important initiative to make assessment of practicals fair, uniform and reliable and to increase the emphasis on practicals in schools, the Board introduced a new scheme of assessment for Class IX from the academic year 2005-06. A document giving detailed guidelines on the new scheme of assessment of practicals with sample question papers was brought out by the Board. This was done to ensure that practicing teachers

understood the new scheme clearly and were sensitized and oriented to the same before it was introduced for the more critical Class X stage. The positive experience and feedback to the new scheme for Class IX convinced the Board that this was a step in the right direction. The Board published another similar document for class X at a later stage. The present document includes questions based on experiments included in class IX as well as class X and has been thoroughly revised.

As per the scheme, theory and practical examination will have a weightage of 60% and 40% respectively. The practical examination will comprise of two components. One component of this practical examination will be in the form of a multiple choice type theory paper test, to be conducted by the Board in Class X as an independent paper. This question paper will be of 20 marks and 1½ hour duration. It will aim at testing of practical skills through multiple-choice type questions. Each multiple-choice question will have four options, with only one of them as the correct option. The second component will also have a weightage of 20 marks but will be conducted at school level on the lines being followed presently.

The Board hopes that this initiative will be an important step not only to give experiments their due place in the subject of Science but also to promote, in general, an experimental culture in our school system.

LIST OF EXPERIMENTS (Only for MCQ Paper)

1. To Prepare

- a) A true solution of common Salt and alum
- b) A suspension of soil, chalk powder and fine sand in water
- c) A colloidal solution of starch in water and egg albumin in water and distinguish between these on the basis of
 - i) Transparency
 - ii) Filtration criterion
 - iii) stability

2. To prepare

- a) a mixture
- b) a compound

using iron filings and sulphur powder and distinguish between these on the basis of :

- i) appearance i.e. homogeneity and heterogeneity
- ii) behaviour towards a magnet
- iii) behaviour towards carbon disulphide a solvant.
- iv) effect of heat.
- 3. To carry out the following chemical reactions and record observations. Also identify the type of reaction involved in each case.
 - i) Iron with copper sulphate solution in water
 - ii) Burning of magnesium in air.
 - iii) Zinc with dilute sulphuric acid.
 - iv) Heating of lead nitrate
 - v) Sodium sulphate with barium chloride in the form of their solutions in water.
- 4. To verify laws of reflection of sound.
- 5. To determine the density of solid (denser than water) by using a spring balance and a measuring cylinder.
- 6. To establish the relation between the loss in weight of a solid when fully immersed in
 - i) tap water
 - ii) strongly salty water, with the weight of water displaced by it by taking at least two different solids.
- 7. To measure the temperature of hot water as it cools and plot a temperature- time graph.

- 8. To determine the velocity of a pulse propagated through a stretched string/slinky.
- 9. To prepare stained temporary mounts of (a) onion peel and (b) human cheek cells and to record observations and draw their labeled diagrams.
- 10. To identify parenchyma and sclerenchyma tissues in plants, striped muscle fibers and nerve cells in animals, from prepared slides and to draw their labeled diagrams.
- 11. To separate the components of a mixture of sand, common salt and ammonium chloride (or camphor) by sublimation.
- 12. To determine the melting point of ice and the boiling point of water.
- 13. To test (a) the presence of starch in the given food samples, (b) the presence of adulterant metanil yellow in dal.
- 14. To study the characteristic of spirogyra/Agaricus, Moss/Fern, Pinus (either with male of female cone) and an Angiospermic plant. Draw and give two identifying features of groups they belong to.
- 15. To observe and draw the given specimens earthworm, cockroach, bony fish and bird. For each specimen, record
 - (a) one specific feature of its phylum
 - (b) one adaptive feature with reference to its habitat.
- 16. To find the pH of the following sample by using pH paper / universal indicator
 - i) Dilute Hydrochloric acid
 - ii) Dilute NaOH solution
 - iii) Dilute Ehanoic acid solution
 - iv) Lemon juice
 - v) Water
 - vi) Dilute sodium Bicarbonate Solution
- 17. To study the properties of acids and bases HCl & NaOH by their reaction with
 - i) Litmus solution (Blue/Red)
 - ii) Zinc metal
 - iii) Solid Sodium Carbonate
- 18. To determine the focal length of
 - i) Concave mirror
 - ii) Convex lensBy obtaining the image of a distant object

- 19. To trace the path of a ray of light passing through a rectangular glass slab for different angles of incidence. Measure the angle of incidence, angle of refraction, angle of emergence and interpret the result.
- 20. To study the dependence of current (I) on the potential difference (V) across a resistor and determine its resistance. Also plot a graph between V and I.
- 21. To determine the equivalent resistance of two resistors when connected in series.
- 22. To determine the equivalent resistance of two resitors when connected in parallel.
- 23. To prepare a temporary mount of a leaf peel to show stomata.
- 24. To show experimetally that light is necessary for photsynthesis.
- 25. To show experimetally that carbon dioxide is given out during respiration.
- 26. To study (a) binary fission in Amoeba and (b) budding in yeast with the help of preapred slides.
- 27. To determine the percentage of water absorbed by raisins.
- 28. To perform and observe the following reactions and classify them into:
 - i) Combination Reaction
- ii) Decomposition Reaction
- iii) Displacement Reaction
- iv) Double Displacement Reaction
- (1) Action of water on quick lime.
- (2) Action of heat on ferrous Sulphate crystals
- (3) Iron nails kept in copper sulphate solution.
- (4) Reaction between sodium sulphate and barium chloride solutions.
- 29. a) To observe the action of Zn, Fe, Cu and Al metals on the following salt solutions.
 - i) $ZnSO_4$ (aq.)
 - ii) FeSO₄ (aq.)
 - iii) CuSO₄ (aq.)
 - iv) Al₂ (SO₄)₃ (aq.)
 - b) Arrange Zn, Fe, Cu and Al metals in the decreasing order of reactivity based on the above result.
- 30. To study the following properties of acid (ethanoic acid):
 - i) odour
 - ii) solubility in water
 - iii) effect on litmus
 - iv) reaction with sodium bicarbonate

CATEGORIES OF PRACTICAL SKILLS

A. Procedural and Manipulative Skills ____

To

- ➤ select appropriate apparatus / instruments for performing the experiment.
- ➤ know the limitations of the apparatus/instruments regarding their size, least count and accuracy.
- ➤ arrange / assemble / set and adjust the apparatus systematically.
- ► handle the apparatus, instruments, chemicals carefully to avoid any damage or injury.
- > perform the experiment with reasonable efficiency and accuracy.
- ➤ separate and remove desired parts of a specimen for detailed study without damaging it.
- use appropriate methods and materials for specimen mounting.
- ▶ locate and rectify the errors in apparatus, instruments, etc.
- ► add chemicals in appropriate quantity.
- ▶ dismantle the experimental set-up carefully.
- practise the precautions in handling sensitive apparatus or chemicals or flame.

B. Observational Skills

To

- ▶ find the least count of the instrument.
- ➤ read the instrument correctly.
- ➤ notice colour change, evolution of gases, formation of precipitates, chemical reactions, etc., carefully.
- ▶ notice the relevant details in the given specimens minutely.
- ▶ locate the desired parts in a specimen accurately.
- ▶ take observations carefully and in a systematic manner.
- ➤ read graph correctly.

C. Drawing Skills

To

- ► make proper observation tables.
- ➤ draw circuit diagrams, ray diagrams, experimental set-ups, sketches, etc. correctly and proportionately.
- ► label sketches and diagrams correctly.
- ▶ draw graphs from observed data correctly.

D. Reporting and Interpretative Skills

To

- ► make a proper plan for recording the observations.
- record the observations/data/information correctly and systematically.
- ► classify and categorize organisms.
- ► make correct calculations/predictions.
- ▶ use proper formulae and mode of summarizing and reporting the result.
- ➤ report the result using correct symbols, units, terms and chemical equations.
- interpret the observations and results correctly.

DESIGN OF THE QUESTION PAPER

Science (Class X) Testing of Practical Skills (Multiple Choice Type Test)

Time allotted: 1½ hours Max. Marks: 20

A. UNIT-WISE WEIGHTAGE

S. No.	UNIT	RELEVANT EXPERIMENTS IN THE SYLLABUS	MARKS ALLOTTED
Class IX	Syllabus		
1.	Food		-
2.	Matter - Its Nature & 1, 2, 3, 11, 12 3 Behaviour 3		3
3.	Organization in the Living world	9,10, 13, 14, 15	3.5
4.	Motion, Force & Work 4, 5, 6, 7, 8 3		3
5.	Our Environment		-
Class X	Syllabus		
6.	Chemical Substances	16, 17, 28, 29, 30	3.5
7.	World of Living	23, 24, 25, 26, 27	3.5
8.	Effect of Current 20, 21, 22 3.5		3.5
9.	Light	18, 19	
10.	Nature Resources	-	-

B. SKILL-WISE WEIGHTAGE

Most questions involve multiple skills and it may not be possible to precisely assign a particular skill to a specific question. The skill-wise weightage given in the table below, may, therefore, be considered as only indicative of what is required in the question paper.

Skill	Weightage (Approximate)
Procedural and Manipulative Skills	35%
Observation Skills	35%
Drawing Skills	15%
Reporting and Interpretative Skills	15%
Total	100%

C. QUESTION-WISE WEIGHTAGE

All the 30 questions are of the multiple choice variety having only one correct answer each. Part A of the question paper contains 20 questions, each carrying 0.5 mark. Part B contains 10 questions, each carrying 1 mark.

S. No.	Estimated difficulty level	Percentage
1.	Easy	15
2.	Average	70
3.	Difficulty	15

D. DIFFICULTY-WISE WEIGHTAGE

E. EXPECTED TIME

Approximate time for reading

and answering one question : 2.5 minutes

Revision time : 15 minutes

Total : 1 hour 30 minutes

Sample Question Paper I

Time: 1 ½ hours
20

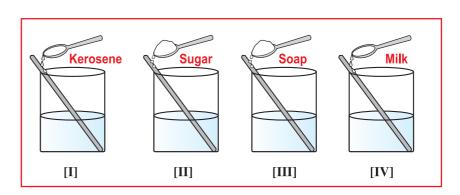
Maximum Marks:

INSTRUCTIONS

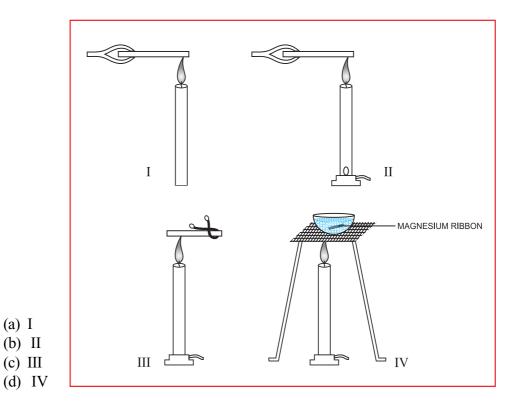
- 1. Attempt all questions.
- 2. There are 30 multiple choice questions in total. Only one of the options in every question is correct.
- 3. The question paper consists of two parts Section A and Section B. Each of the 20 questions in Section A carries 0.5 mark and each of the 10 questions in Section B carries 1.0 mark.

SECTION - A

- 1. The following substances are added to water in a beaker as shown below. The mixture is stirred well. A true solution is found in the beaker
 - (a) I (b) II (c) III (d) IV



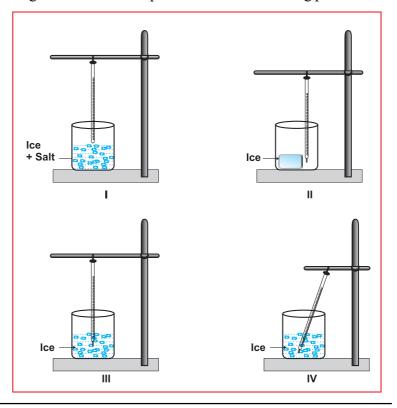
- 2. When we start heating a mixture of sulphur powder and iron fillings, we would observe that
 - (a) surphur starts melting.
 - (b) iron filings start melting.
 - (c) mixture becomes red hot.
 - (d) mixture evaporates
- 3. Four students used different ways of burning magnesium ribbon during an experiment as shown below. The correct way has been followed by student:



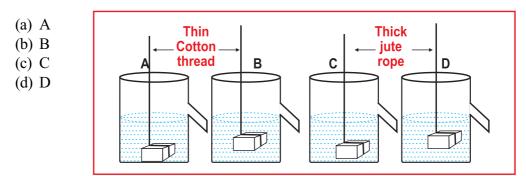
- 4. Which one of the following is the correct set up to determine the melting point of ice?
 - (a) I

(a) I

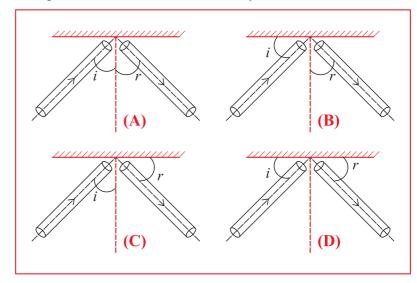
- (b) II
- (c) III
- (d) IV



5. In the experiment to establish the relation between loss in weight of an immersed solid with the weight of water displaced by it, the correct setup is shown in figure



- 6. While doing experiment on verifying the law of reflection of sound, four students measured the angles < i and <r as shown in the diagram below. The correct measurement of the angles of incidence and angle of reflections, has been done by student.
 - (a) A (b) B
 - (c) C
 - (d) D

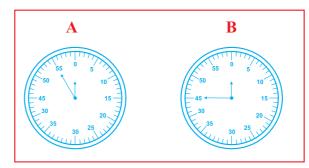


- 7. The table below gives the observations reported by two students X and Y for an experiment on the study of temperature-time graph. The experiment is likely to have been performed correctly by
 - (a) X.
 - (b) Y.
 - (c) both X and Y.
 - (d) neither X nor Y.

Time	Temp (°C)	observed by
(min)	Student X	Student Y
0	61.0	61.0
2	60.5	59.0

4	60.0	58.0
6	59.0	57.5
8	58.0	57.0
10	56.5	56.5
12	54.0	56.0

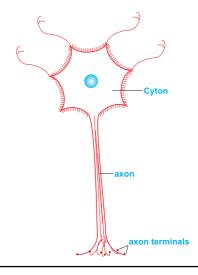
- 8. A strong transverse horizontal pulse, created at one end of a string, is observed to complete 5 single journeys (from one end to other end) along its length, before fading out. The initial and final readings, on a stop-clock used in the experiment, are as shown here. If the length of the string is L metre, the speed of the pulse, through the string, is
 - (a) (L/9) ms¹
 - (b) (L/10) ms¹
 - (c) (L/45) ms¹
 - (d) (L/50) ms¹



- 9. Nikita observed a slide of cheek cells, first under low power and then under high power of a microscope. Under the low power she must have seen
 - (a) fewer cells in a darker field of view.
 - (b) more cells in a brighter field of view.
 - (c) more cells in a darker field of view.
 - (d) fewer cells in a brighter field of view.
- 10. Raj observed nerve cells under the microscope and made the following sketch.

The mistake in his drawing is that cyton is shown to have

- (a) cilia
- (b) dendrites
- (c) nucleus
- (d) cytoplasm.



- 11. A student added 2 drops of iodine solution into 4 ml of starch solution in test tube A, another student added 2 ml of starch solution into 4 ml of iodine solution in test tube B. They would then observe
 - (a) a change of colour to blue black in test tube A but not in test tube B.
 - (b) a change of colour to blue black in test tube B but not in test tube A.
 - (c) a change of colour to blue black in both test tubes A and B
 - (d) no change of colour in any test tube.
- 12. A student observed that the colour of pH paper changed to green when she dipped it in water. She added a few drops of concentrated hydrochloric acid to the water. The colour of pH would turn to
 - (a) light red.
 - (b) apple green.
 - (c) dark blue.
 - (d) lemon yellow.
- 13. 10 mL of freshly prepared iron sulphate was taken in each of four test tubes. Strips of copper, iron, zinc and aluminium were introduced, each metal in a different test tube. A black residue was obtained in two of them. The right pair of metals forming the precipitates is
 - (a) copper and zinc.
 - (b) aluminium and copper.
 - (c) iron and aluminium.
 - (d) zinc and aluminium.
- 14. The following symbols are usually shown on the bottles of commercial acetic acid.

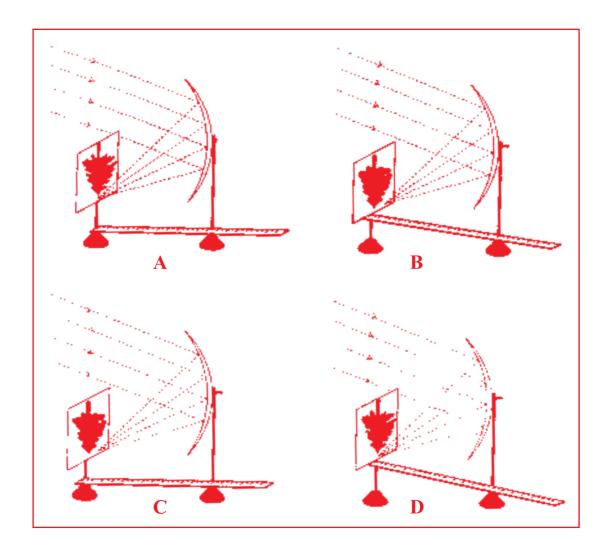




The symbols indicate that acetic acid is

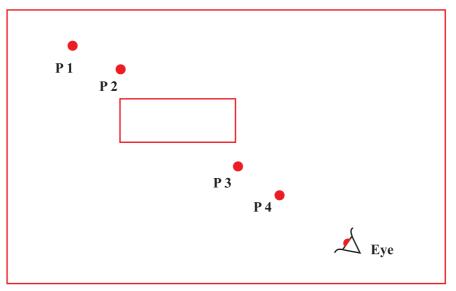
- (a) corrosive and flammable.
- (b) radioactive and flammable.
- (c) oxidizing and corrosive.
- (d) flammable and explosive.

15. Four student A, B, C and D carried out measurement of focal length of a concave mirror as shown in the four diagrams.



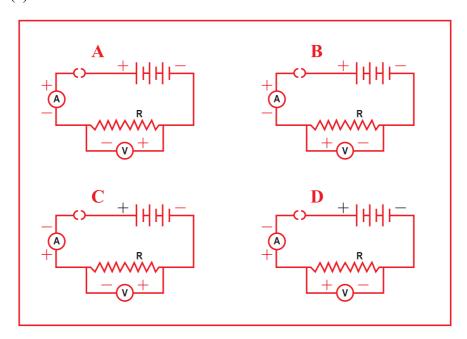
The best result will be obtained by student

- (a) A
- (b) B
- (c) C
- (d) D
- 16. In the glas slab experiment shown below, four students A, B, C and D did the following:
 - A: kept the eyes far from the glass slab while placing both the pins P_3 and P_4
 - B: kept the eyes close to the glas slab while placing both the pins P_3 and P_4
 - C: kept the eyes close to the glass slab while placing pin P₃ and and far from the slab
 - while placing pin P₄
 D: kept the eyes far from the glass slab while placing pin P₃ and close to the slab while placing pin P₄

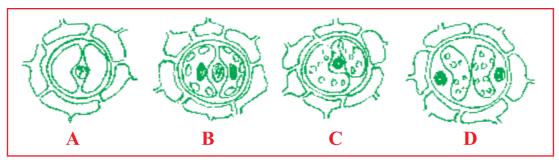


The correct procedure is that of student

- (a) A.
- (b) B.
- (c) C.
- (d) D.
- 17. Out of the four circuits shown for studying the dependance of the current on the potential difference across a resistor, the correct circuit is
 - (a) A
 - (b) B
 - (c) D
 - (d) E



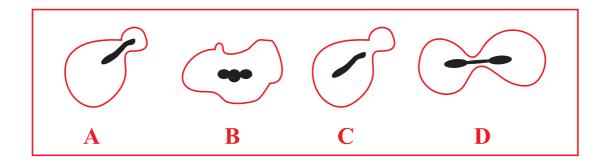
18. Students observed the epidermal peel of a leaf under the high power of a microscope. The following are the sketches made by them.



The correct sketch is

- (a) A
- (b) B
- (c) C
- (d) D
- 19. Student A, B and C were given five raisins each of equal weight. The raisins were soaked in distilled water at room temperature. A removed the raisins after 30 minutes, B after two hours and C after one hour. If P_A , P_B and P_C denote percentage absorption of water obtained by students A, B and C respectively, then

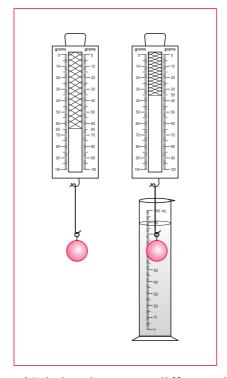
 - (a) $P_A > P_B > P_C$ (b) $P_A < P_B < P_C$ (c) $P_A < P_B > P_C$ (d) $P_A = P_B = P_C$
- 20. Out of the following diagrams which one depicts a stage in binary fission of amoeba.



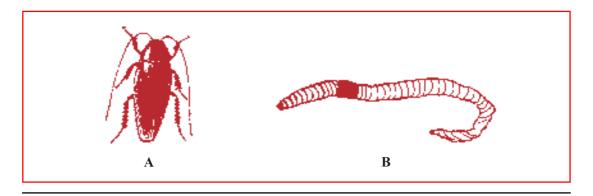
- (a) A
- (b) B
- (c) C
- (d) D

SECTION - B

- 21. Which one of the following would be the correct set of apparatus required if you have to separate camphor and common salt?
 - (a) Round bottom flask, funnel, burner, condensor, wire guage, stand with clamp
 - (b) Conical flask, filter paper, funnel, beaker, stand with clamp, wire guage
 - (c) Separating funnel, beaker, conical flask tripod stand, burner wire guaze
 - (d) China dish, funnel, burner, cotton plug, tripod stand, stand with clamp, wire guaze.
- 22. A student notes down the observations in the two spring balances and the measuring cylinder shown in the figure. From the given observations, the volume of the solid is
 - (a) 64 cc
 - (b) 36 cc
 - (c) 30 cc
 - (d) 100 cc



23. Given below are pictures of two animals A and B belonging to two different phyla. Which characteristic features of their bodies are specific features of thier respective phyla?



- (a) Antennae of A and segments of B
- (b) Three pairs of appendages of A and clitellum of B
- (c) Jointed appendages of A and segmented body without appendages
- (d) Wings of A and slender body of B
- 24. The teacher had shown the students two specimens A and B from the plant kingdom whose pictures are given below. Which was the correct identification and classification into its group?

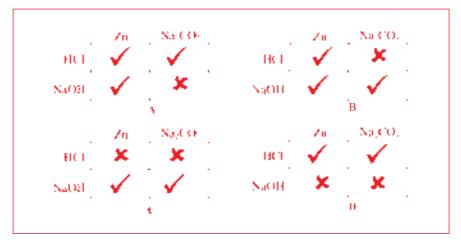


В

- (a) Cone of gymnosperm and flower of angiosperm.
- (b) Cone of angrosperm and flower of gymnosperm.
- (c) Cone and flower of an angiosperm,
- (d) Cone and flower of a gymnosperm.
- 25. Student were asked to study the reaction between barium chloride and sodium sulphate. Four different reports of the experiment are given below. Choose the correct one.

	Procedure	Observation
a	Mixed powder of barium chloride and sodium sulphate.	The colour of the mixture changes to yellow.
b	Mixed solution of barium chloride to sodium sulphate.	Thick white precipitate is formed.
С	Added solution of barium chloride to sodium suphate powder	Solution become turbid
d	Added powder of barium chloride to sodium sulphate solution.	No change is observed.

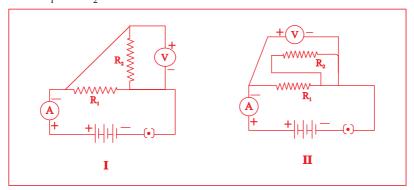
26. Four students studied reactions of zinc and sodium carbonate with dilute hydrochloric acid and dilute sodium hydroxide solutions and presented their results as follows. The ' $\sqrt{}$ ' represents evolution of gas, whereas 'X' represents absence of any reaction.



The right set of observations is that of student

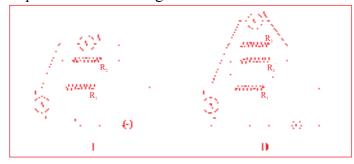
- (a) A
- (b) B
- (c) C
- (d) D

27. The resistors R_1 and R_2 are connected in

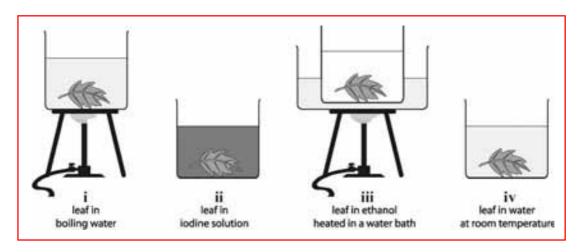


- (a) parrellel in both circuits.
- (b) series in both ciruits.
- (c) parrallel in circuit I and in series in ciruit II.
- (d) series in circuit I and in parallel in cicuit II.
- 28. Circuit I: ammeter reads current i_1 , and voltmeter reads V_1 Circuit II: ammeter reads current i_2 and voltmeter reads V_2

The relationship between the readings is



- (a) $i_1 > i_2$,; $V_1 = V_2$ (b) $i_1 < i_2$,; $V_1 = V_2$ (c) $i_1 > i_2$,; $V_1 > V_2$ (d) $i_1 < i_2$,; $V_1 < V_2$
- 29. A student performed the starch test on a leaf. Some steps involved are shown below:

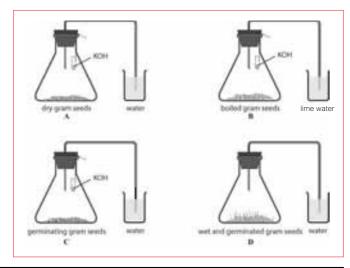


The correct swquence of steps should be

- (a) iv; iii; ii; i
- (b) i; ii; iii; iv
- (c) ii; iii; iv; i
- (d) i; iii; iv; ii
- 30. After performing the experiment to show that germinating seeds give out carbon dioxide during respiration, students drew the following diagrams.

The correct labelled diagram is

- (a) A
- (b) B
- (c) C
- (d) D



Key for Sample Paper 1

Q. No.	Key	Explanation
1.	(b)	Sugar makes a true solution.
2.	(a)	Sulphur has a lower melting point than iron.
3.	(c)	Magnesium ribbon is to be burnt directly on a bunsen burner using a pair of tongs.
4.	(c)	Only in this set up, the bulb is inside the crushed ice and thermometer is straight to give the correct melting point.
5.	(b)	The solid must be suspended by an inextensible string in the centre of the overflow can without touching its bottom.
6.	(a)	The angles of incidence and reflection are measured with respect to the normal at the point of incidence.
7.	(b)	The rate of fall of temperature is faster first and slower later.
8.	(b)	The time taken for one journey is $(5+45)/5$ second i.e. 10 second Hence the speed of the pulse is $(L/10)$ ms ⁻¹
9.	(b)	Lower the magnifying power, more is the number of cells seen. (in a brighter field.)
10.	(a)	No cilia in cyton.
11.	(c)	Iodine + starch solution → Blue - black colour.
12.	(a)	The colour of pH paper is green in neutral medium, whereas it is red in acidic medium.
13.	(d)	Zinc and aluminium being more reactive will replace iron from iron sulpate.
14.	(a)	Acetic acid is corrosive and flammable.
15.	(a)	For the correct measurement of focal length, must have a sharp image on the screen and the meter scale must be correctly positioned between the (sharp image) screen and the centre of the concave mirror.
16.	(a)	We need to keep the eye far from the glass slab to have a good and proper alignment of the pins.

- 17. (b) We must not only put the ammeter in series and the voltmeter in parallel (with the resistor) but also ensure that the polarities of both the instruments are correct.
 18. (b) Guard cells have nucleus as well as chloroplasts.
 19. (c) Absorption of water increases with time up to its maximum limit.
- 20. (d) Shows binary fission in amoeba
- 21. (d) Only in this set all necessary appraratus is given to set up an experiment for sublimation.
- 22. (c) The volume of the solid (in cc) has the same magnitude as its loss in weight (in grams) in water.
- 23. (c) Typical characteristics of phylum arthropoda and annelida
- 24. (a) Corret identification and clasification.
- 25. (b) It is a double displacement reaction in which exchange of icons takes place in aqueous solution only.
- 26. (a) Zinc reacts with dilute HCl and NaOH, whereas Na₂CO₃ reacts only with dilute HCl.
- 27. (c) We must not look for a stereotyped circuit diagram but look for the basic condition for (i) parallel (ii) series connection of two resistors in a given circuit.
- 28. (b) The equivalent resistance, of a parallel combination of resistors, is less than the resistance of either of its two branches. The equivalent resistance, in circuit 2, is, therfore, less than (R_1+R_2) (the equivalent resistance of circuit 1) and hence the current flowing through it increases. The voltage reading, in both cases, is, however, the same.
- 29. (d) Boiling kills the cells, chlorophyll leaches out when boiled in ethanol, but the leaf becomes brittle, made normal by washing it in water. Starch gets stained with iodine.
- 30. (c) Labelling is correct for seeds in the conical flask, KOH in the suspended test tube and water in the beaker.

Sample Question Paper II

Time: 1 ½ hours Maximum Marks: 20

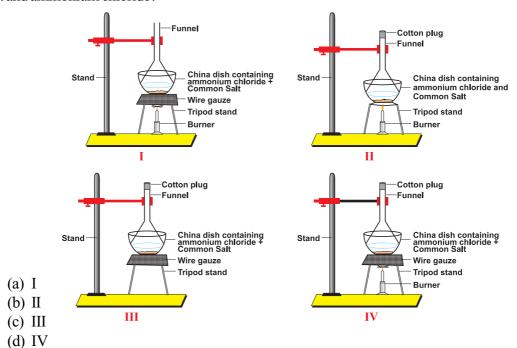
INSTRUCTIONS

- 1. Attempt all questions
- 2. There are 30 multiple choice questions in total. Only one of the options in every question is correct.
- 3. The question paper consists of two parts Section A and Section B. Each of the 20 questions in Section A carries 0.5 mark and each of the 10 questions in Section B carries 1.0 mark.

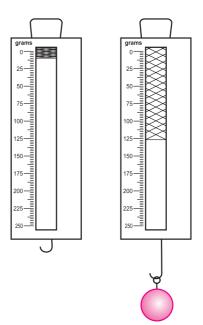
SECTION - A

- 1. A student was asked to mix the white of an egg with water and stir well. The student observed that
 - (a) a transparent solution is formed.
 - (b) a translucent mixture is formed.
 - (c) egg white settles down at the bottom.
 - (d) egg white floats on the surface of the water.
- 2. A student by mistake mixed iron fillings and sulphur powder. He wanted to separate them from each other. The method you would advise him to use is dissolve the mixture in
 - (a) boiling water.
 - (b) cold water.
 - (c) carbon disulphide.
 - (d) kerosene.
- 3. The colour of NO₂ and PbO formed when lead nitrate is heated are:
 - (a) brown and green
 - (b) brown and yellow
 - (c) brown and colourless
 - (d) colourless and yellow

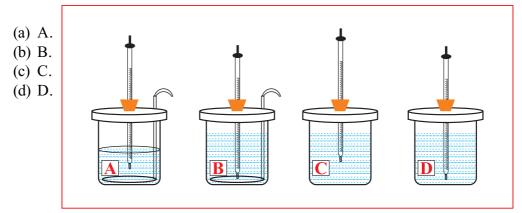
4. Which of the following is the correct arrangement for separation of mixture of common salt and ammonium chloride?



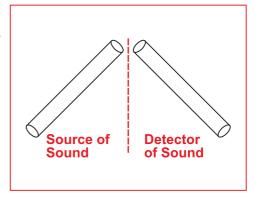
- 5. The spring balance shown here is used to measure the mass of given solid. The mass of the solid is
 - (a) 115 g
 - (b) 118 g
 - (c) 120 g
 - (d) 125 g



6. Of the four experimental set-ups shown here to study temperature-time graph, the best one is



- 7. For doing his experiment on verifying the laws of reflection of sound, a student sets up his apparatus as shown. The experiment is more likely is get performed successfully if the screen shown is a
 - (a) plane wooden board.
 - (b) wooden board with many holes in it.
 - (c) a foam padded board.
 - (d) a sheet of pure white cloth.



8. Four students did their experiment on measuring the speed of a pulse through a string as follows:

Student A stretched his thick cotton string very taut and gave it a very mild transverse horizontal jerk

Student B stretched his thin jute string just taut and gave it a mild transverse horizontal jerk

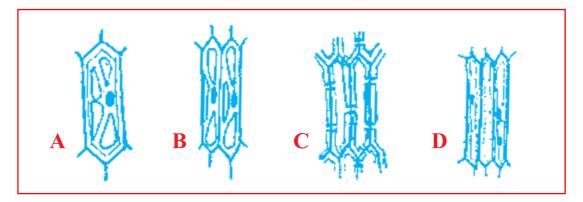
Student C stretched his thick cotton string just taut and gave a strong transverse horizontal jerk

Student D stretched his thin jute string very taut and gave it a strong transverse horizontal jerk.

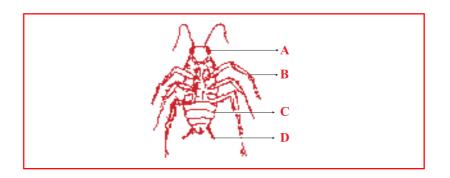
The best choice is that of student

- (a) A
- (b) B
- (c) C
- (d) D

- 9. While observing a thin section of a plant stem, four students sketched sclerenchyma as given below. The correct diagram is
 - (a) A.
 - (b) B.
 - (c) C.
 - (d) D.

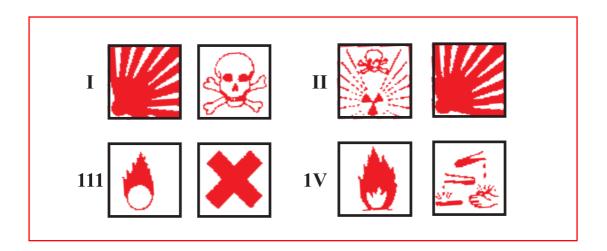


- 10. A student wants to observe a spiracle of a cockroach. She should observe the region on its body shown by the label
 - (a) A.
 - (b) B.
 - (c) C.
 - (d) D.



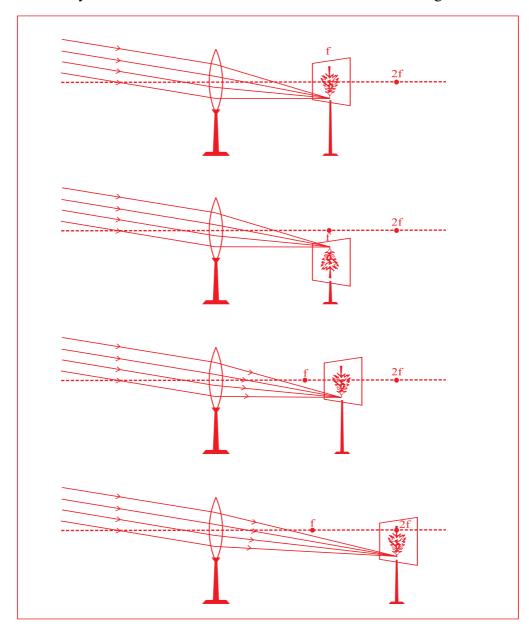
- 11. To observe starch granules in potato under a microscope, freshly cut surface of potato was pressed on a slide. The stain that will show starch granules clearly is
 - (a) methylene blue
 - (b) iodine
 - (c) safranine
 - (d) eosine

- 12. Bottle A contains oxalic acid and bottle B contains sodium carbonate solution. When pH paper is dipped in each of the solutions, the colour seen in A and B respectively be
 - (a) orange, blue.
 - (b) blue, orange.
 - (c) green, blue.
 - (d) orange, green.
- 13. The pair of safety symbols you notice on the bottles of commercial acetic acid available in the laboratory, is shown in
 - (a) I.
 - (b) II.
 - (c) III.
 - (d) IV



- 14. When you place iron nail in copper sulphate solution, the reddish brown coating formed on the nail is
 - (a) soft and dull.
 - (b) hard and flaky.
 - (c) smooth and shining.
 - (d) rough and granular.

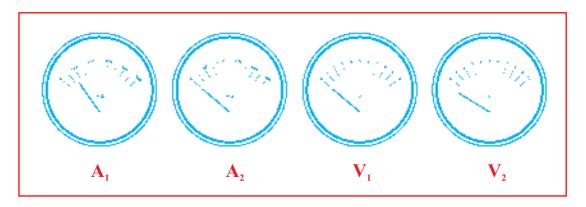
15. Parallel rays from a distant tree incident on a convex lens form an image on the screen.



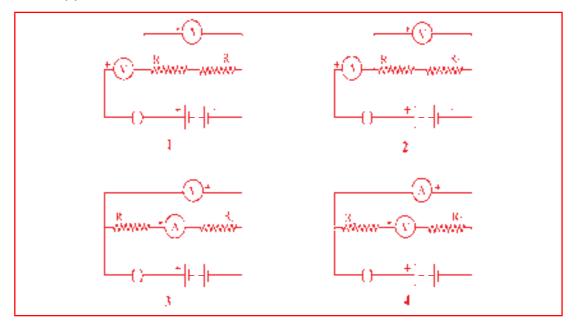
The diagram correctly showing the image of the tree on the screen is

- (a) A.
- (b) B.
- (c) C.
- (d) D.
- 16. The positions of the pointers of the two ammeters A_1 and A_2 , and two voltmeters V_1 and V_2 available in the laboratory are shown in the given figure. For an experiment to study the dependence of the current on the potential difference across a resistor, the student would prefer

- (a) ammeter A_1 and voltmeter V_1 .
- (b) ammeter A_2 and voltmeter V_1 .
- (c) ammeter A_1 and voltmeter V_2 .
- (d) ammeter A_2 and voltmeter V_2 .



- 17. The correct way of connecting the ammeter and voltmeter with a series combination of two resistors in a cicuit for finding their equivalent resistance, is shown in diagram
 - (a) 1.
 - (b) 2.
 - (c) 3.
 - (d) 4.

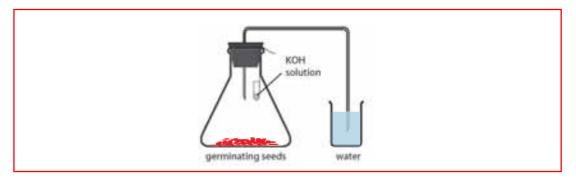


18. While preparing a temporary mount of stomata, four students used different stains given below

STUDENT	STAIN
A	Acetocarmine
В	Methylene Blue
С	Safranine
D	Iodine

The correct stain was used by the student

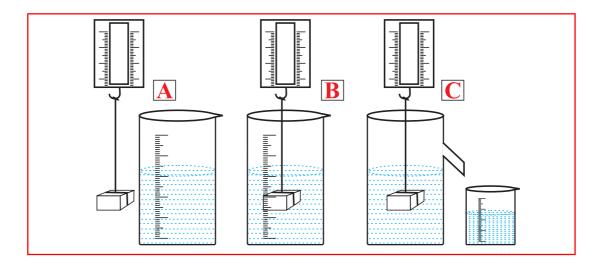
- (a) A
- (b) B
- (c) C
- (d) D
- 19. A part of de-strached leaf of a potted plant was covered with black paper strips on both sides and the plant was kept in sunlight for 8 hours. The leaf was then tested with iodine after boiling it in alcohol. Only the uncovered part of the leaf turned blue black. The inference is that
 - (a) CO₂ is necessary for photosynthesis.
 - (b) light is necessary for photosynthesis.
 - (c) chlorophyll is necessary for photosynthesis.
 - (d) water is necessary for photosynthesis.
- 20. In the experiment shown in the figure, water is found to rise in the bent tube. The reason is that



- (a) CO₂ in the bent tube dissolves in water.
- (b) Partial vacuum is created in the flask.
- (c) Atmospheric pressure outside is reduced
- (d) Water in the bent tube rises due to capillary action.

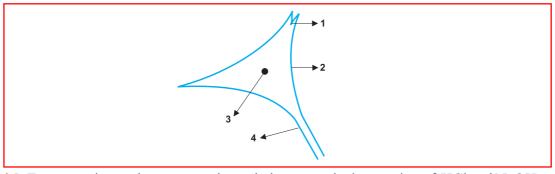
SECTION - B

- 21. Which of the following set of apparatus is required to determine the boiling point of water?
 - (a) tripod stand, conical flask, thermometer, wire guaze, stand with clamp, pair of tongs
 - (b) funnel, burner, test tube, thermometer, wire guaze, stand with clamp
 - (c) boiling tube, beaker, thermometer, burner, cork with one hole, stand with clamp, wire guaze
 - (d) round bottom flask, burner, thermometer, wire guaze, stand with clamp, cork with 2 holes, glass tube
- 22. The readings of the spring balance will be
 - (a) equal to each other in all cases A, B and C.
 - (b) equal to each other in cases A and C only.
 - (c) equal to each other in cases B and C only.
 - (d) different in every case.

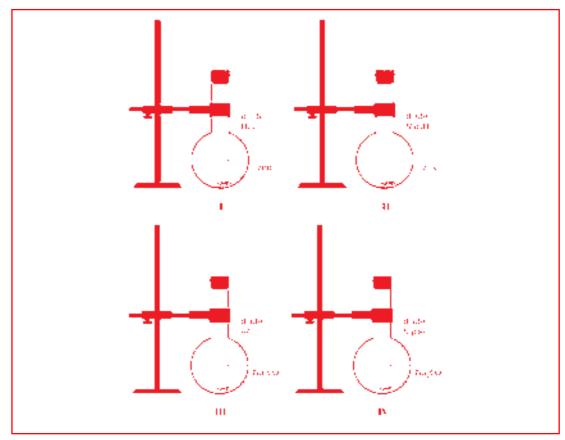


- 23. You are viewing a prepared slide of striped muscle fibers from cockroach leg. When you focus the microscope, the striations appear pale and indistinct. To make the striations clearly visible, you would
 - (a) slowly close the diaphragm to reduce the light.
 - (b) remove the mirror to cut out light.
 - (c) change the eye piece to increase magnification.
 - (d) replace the objective to decrease magnification.

- 24. A figure depicting parts of a neuron is given below. The correct identification of the labels 1, 2, 3, 4 respectively is
 - (a) dendrite, cytoplasm, Nissl granules, nerve fibre.
 - (b) cilla, endoplasmic reticulum, nucleoli, nerve fibre.
 - (c) dendrons, cell body, Nissl granules, axon.
 - (d) dendrites, cyton, nucleus, axon.



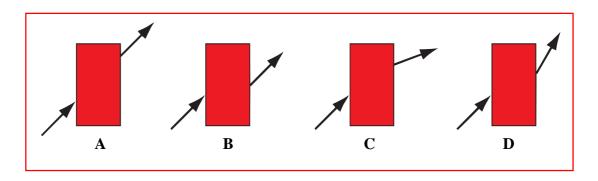
25. Four experimental set ups are shown below to study the reaction of HCl and NaOH on zinc and sodium carbonate.



The set ups that would result in a rapid evolution of gas would be

- (a) I and III
- (b) II and IV
- (c) I and II
- (d) III and IV

- 26. Which of the following is a correct observation when water is added to lime?
 - (a) No change and a hissing sound
 - (b) Vigorous bubbling and a hissing sound
 - (c) Slow bubbling with no sound
 - (d) Vigorous bubbling with evolution of heat and a hissing sound
- 27. Four students showed the following traces of the path of a ray of light passing through a rectangular glass slab.



The trace most likely to be correct is that of student

- (a) A.
- (b) B.
- (c) C.
- (d) D.
- 28. The following apparatus is available in a laboratory.

Cell : adjustable from 0 to 1.5 volt

Resistor : 4Ω and 12Ω

Ammeters : A_1 of Range 0 to 3 A : Least Count 0.1 A

A₂ of Range 0 to 1 A: Least Count 0.05 A

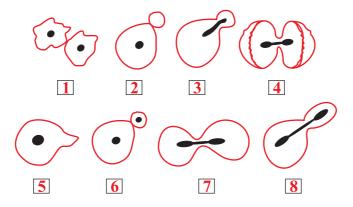
Voltmeters: V_1 of Range 0 to 10 V: Least Count 0.5 V

V₂ of Range 0 to 5 V : Least Count 0.1 V

The best combination of voltmeter and ammeter for finding the equivalent resistance of the resistors in parallel would be

- (a) ammeter A_1 and voltmeter V_1 .
- (b) ammeter A_1 and voltmeter V_2 .
- (c) ammeter A_2 and voltmeter V_1 .
- (d) ammeter A_2 and voltmeter V_2 .

29. From the following diagrams, select the correct ones showing stages of binary fission in amoeba and budding in yeast in their proper sequence



- (a) 5, 1, 4 and 2, 3, 6.
- (b) 3, 4, 7 and 2, 8, 6.
- (c) 7, 4, 1 and 3, 8, 6.
- (d) 8, 7, 4 and 3, 2, 6.
- 30. Student weighed 108 raisins and designated the weight as A. She then soaked them in 50 mL distilled water in a beaker. After 2 hours, she removed the raisins wiped them dry from outside and wighed again and called that weight as B. For determining the percentage of water obsorbed by raisins, she should calculate as follows:
 - (a) $\frac{B-A}{A} \times 100$
 - (b) $\frac{B-A}{B} \times 100$
 - (c) $\frac{B-A}{A} \times \frac{1}{100}$
 - (d) (B-A) x 100

Key for Sample Paper II

Q. No.	Correct Choice	Explanation/Remarks
1.	(b)	Because the egg white forms colloids
2.	(c)	Sulphur is souble in carbon disulphide
3.	(b)	NO ₂ is brown and PbO is yellow.
4.	(d)	This is correct arrangement because of presence of burner, wire gauze and cotton plug.
5.	(a)	The zero error has to be correctly accounted for.
6.	(a)	We need a stirrer and a thermometer correctly positoned without touching the sides of the colorimeter.
7.	(a)	All the other screens would absorb/transmit sound.
8.	(c)	String should not be stretched too taut and the jerk should be as strong as possible.
9.	(c)	Pits, non-nucleated cells and thick cell walls are the characteristics of sclerenchyma.
10.	(c)	Correct location of spiracles.
11.	(b)	Test for starch
12.	(a)	The colour of pH paper is orange in acidic medium while it is blue in basic medium.
13.	(d)	Acetic acid is flammable and corrosive.
14.	(a)	The freshly deposited copper is soft and dull.
15.	(a)	The image of the (vertical) tree on the screen will be an inverted vertical image at f.
16.	(b)	We should select instruments without any zero error.
17.	(b)	The ammeter must be connected in series, between the battery and the series combination of the two resistors, and the voltmeter should be put in parallel across the series combination of the two resistors. All the polarities must also be correct.
18.	(c)	Temporary mount of stomata is prepared with safranine.
19.	(b)	The covered part did not get sunlight which is necessary for photosynthesis.

- 20. (c) Temporary nount of stornata is prepared with safranine.
- 21. (d) This set of apparatus contains all the required items for determination of boiling point.
- 22. (c) The loss in weight of a solid does not depend on the shape of the vessel or the volume of water in it.
- 23. (a) Reduction of light gives better contrast.
- 24. (d) Correct labelling.
- 25. (a) The reactions between (i) Zinc and dil. HCl and (ii) dil HCl and Na₂CO₃ are fast whereas the reaction between Zn and NaOH is slow. There is no reaction between NaOH and Na₂CO₃
- 26. (d) The combination reaction between line (CaO) and water is a vigrous, exothermic and produces hissing sound.
- 27. (b) The emergent ray, from the retangular glass slab, is parallel to the incident ray and is laterally displaced to the left (or lower side) of the incident ray.
- 28. (d) The overall range of the voltage is from 0 to 1.5V and that of current is from 0 to 1.5/3 A=0.5A. We, therefore, perfer, instruments that cover these ranges and also have a better least count.
- 29. (c) Correct sequence of binary fission in amoeba and budding in yeast.
- 30. (a) Correct formula

Sample Question Paper III

Time: 1 ½ hours Maximum Marks: 20

INSTRUCTIONS

- 1. Attempt all questions
- 2. There are 30 multiple choice questions in total. Only one of the options in every question is correct.
- 3. The question paper consists of two parts Section A and Section B. Each of the 20 questions in Section A carries 0.5 mark and each of the 10 questions in Section B carries 1.0 mark.

SECTIONA

- 1. The correct order which describes the true solution, colloidal solution and suspension in the order of their increasing stability is:
 - (a) Suspension < colloidal solution < true solution
 - (b) Colloidal solution < true solution < suspension
 - (c) True solution < colloidal solution < suspension
 - (d) Colloidal solution < suspension < true solution
- 2. To prepare iron sulphide, by heating a mixture of iron filings and sulphur powder, we should use a
 - (a) test tube.
 - (b) watch glass.
 - (c) china dish.
 - (d) petri dish.
- 3. When a mixture of common salt and ammonium chloride is heated, it is observed that
 - (a) Solid common salt gets deposited on the cooler parts of the funnel while solid ammonium chloride remains in the china dish.
 - (b) Mixture of common salt and ammonium chloride turns into greenish crystals when allowed to cool.
 - (c) Ammonium chloride gets deposited on the cooler parts of the funnel and solid common salt remains in the china dish.
 - (d) droplets containing both common salt and ammonium chloride appear on the upper part of the funnel while some molten mixture of common salt and ammonium chloride remains in the china dish.

- 4. Copper sulphate solution is added to a test tube containing a clean iron nail. The correct description regarding the deposition of copper on the iron nail would be that it starts depositing
 - (a) at the tip of the nail
 - (b) from the head of the nail
 - (c) in the middle of the nail
 - (d) anywhere on the nail
- 5. While determining the density of a copper piece using a spring balance and a measuring cylinder, Seema carried out the following procedure:
 - i. noted the water level in the measuring cylinder without the copper piece
 - ii. immersed the copper piece in the water
 - iii. noted the water level in the measuring cylinder with the copper piece inside it.
 - iv. removed the copper piece from the water and immediately weighed it using a spring balance.

The wrong step in the procedure is

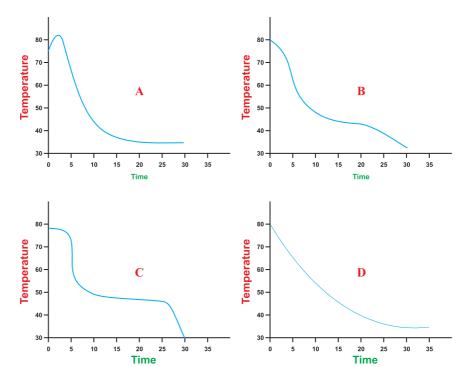
- (a) step (i)
- (b) step (ii)
- (c) step (iii)
- (d) step (iv)
- 6. The temperature-time variation graphs, as obtained by four students A, B, C and D are as shown. The graph, likely to be correct is that of student.







(d) D.



- 7. A laboratory had the following apparatus available in it:
 - (A) Two thin hollow wooden tubes
 - (B) An intense & broad source of sound
 - (C) An Intense, pointed source of sound
 - (D) A sharp pointed detector to detect the sound
 - (E) A well polished metal sheet
 - (F) A white painted thermocol sheet

A student can do his experiment to verify the laws of reflection of sound successfully by choosing the appartus labelled as

- (a) A, C, D, E
- (b) A, C, E
- (c) A, B, D, E
- (d) A, C & D
- 8. While doing the experiment on measuring the velocity of a pulse through a stretched string, a student had to choose between a
 - (i) thick silk string and a thick cotton string
 - (ii) stop clock and a table clock.

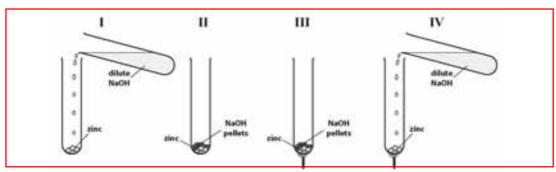
The combination choice that he should prefer is

- (a) silk string and the table clock
- (b) silk string and the stop clock
- (c) cotton string and the table clock
- (d) cotton string and the stop clock
- 9. On adding water to a sample of arhar dal (tuvar dal) a student found that the water became yellow in colour. He took a sample of this yellow water and added a few drops of dilute HCl. The sample became pink. This confirms that the adulterant added to the dal was
 - (a) Turmeric
 - (b) metanil yellow
 - (c) potassium dichromate
 - (d) yellow dye

- 10. Given below are four operations for preparing a temporary mount of human cheek cells:
 - (i) taking scraping from inner side of the cheek and spreading it on a clean slide
 - (ii) putting a drop of gycerine on the material
 - (iii) adding two or three drops of methylene blue
 - (iv) rinsing the mouth with fresh water and disinfectant solution.

The correct sequence of these operations is

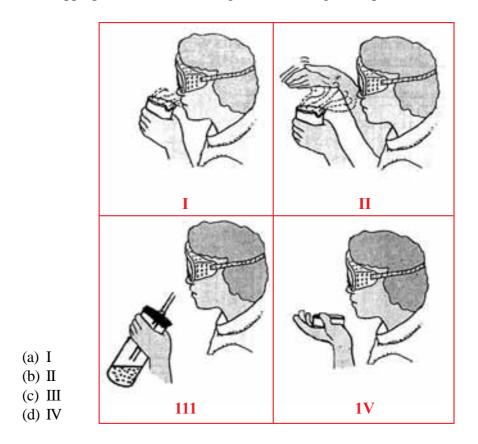
- (a) i-ii-iii-iv
- (b) iv-i-iii-ii
- (c) iv-i-ii-iii
- (d) i-iii-ii-iv
- 11. The specimens of a bird and a fish were given to students to identify the feature / features that places them in the same phylum. The correct identifying feature / features would be
 - (a) pointed heads.
 - (b) bulky thorax.
 - (c) presence of scales.
 - (d) post anal tails.
- 12. The figures below show set-ups for studying the reaction of zinc with sodium bydroxide.



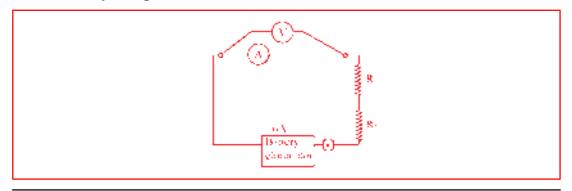
The correct set-up is

- (a) I.
- (b) II.
- (c) III.
- (d) IV.
- 13. Two beakers A and B contain Iron (II) sulphate solution. In the beaker A is placed a small piece of copper and in the beaker B is placed a small piece of zinc. It is found that a grey deposit forms on the zinc but not on the copper. From these observations it can be concluded.
 - (a) Zinc is most active metal followed by iron and copper.
 - (b) Zinc is most active metal followed by copper and then iron.
 - (c) Iron is most active metal followed by zinc and then copper.
 - (d) Iron is most active metal followed by copper and then zinc.

14. The most appropriate method of testing the odour of a given liquid is



- 15. In an experiment to determine the focal length of a convex lens, a student obtained a sharp inverted image of distant tree on the screen behind the lens. She then removed the screen and looked through the lens in the direction of the object. She will see
 - (a) an inverted image of the tree at the focus of the lens.
 - (b) no image as the screen has been removed.
 - (c) a blurred image on the wall of the laboratory.
 - (d) an erect image of the tree on the lens.
- 16. In an experiment, to find the equivalent resistance of a series of combination of two resistors R_1 and R_2 a student uses the ciruit shown here.



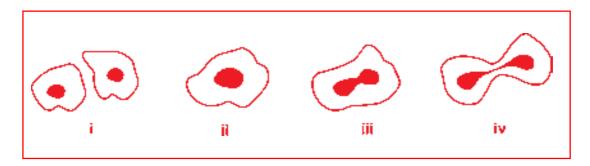
The circuit will give

- (a) correct reading for volage V, but incorrect reading for current I.
- (b) correct reading for current I, but incorrect reading for voltage V.
- (c) correct reading for both current I and voltage V.
- (d) incorrect readings for both current I and voltage V.
- 17. In an experimet to study dependence of current on the potential difference across a given resistor, four students P, Q, R and S kept the plug key in the circuit closed for time t₁ and then open for time t₂ as given in the table below.

Student	Closed time $t_{_1}$ seconds	Open time t_2 seconds
P	30	60
Q	60	30
R	60	15
S	45	15

The best choice of open and closed time is that of student

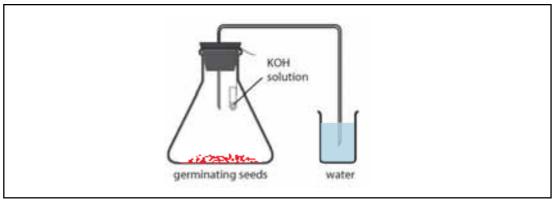
- (a) P.
- (b) Q.
- (c) R.
- (d) S.
- 18. The following figures illustrate binary fission in Amoeba in an incorrect sequence.



The correct sequence is

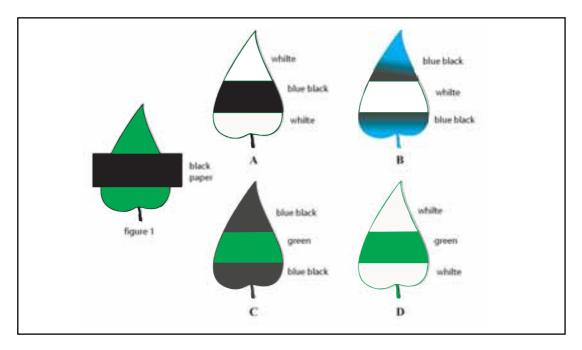
- (a) (i), (iii), (iv), (ii)
- (b) (ii), (iii), (iv), (i)
- (c) (iv), (iii), (ii), (i)
- (d) (iii), (iv), (ii), (i)

19. The following experiment is set up to show that a gas is released during respiration by germinating seeds.



In this set up, the small test tube containing KOH solution is kept in the conical flask to absorb

- (a) air in the flask.
- (b) moisture in the flask.
- (c) O_2 from the air in the flask.
- (d) \tilde{CO}_2 released by the germinating seeds.
- 20. A leaf from a de-strached plant is covered with black paper strip as shown in figure 1. The starch test is done on the leaf after 8 hours.

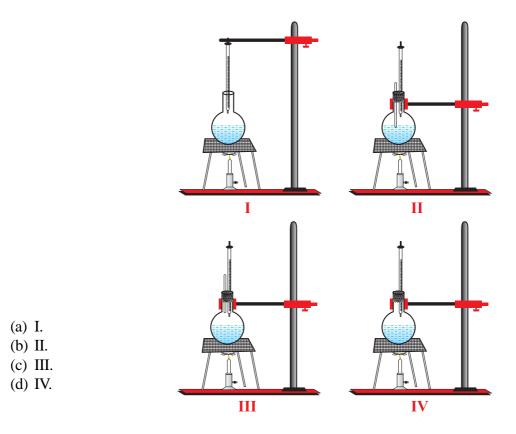


The results will be as shown in diagram

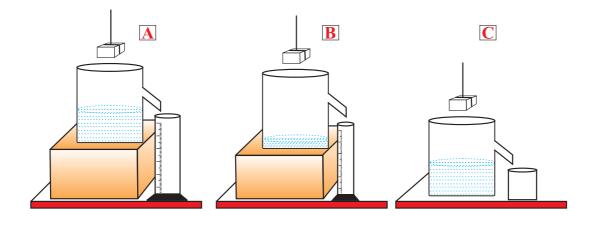
- (a) A.
- (b) B.
- (c) C.
- (d) D.

SECTION B

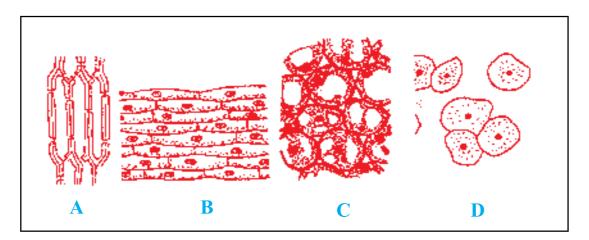
21. Which one of the following experimental arrangements is correct for the determination of boiling point of water?



- 22. Three students A, B and C determined the volume of a solid by immersing it in water in the overflow cans set up as shown. The result obtained will be wrong for
 - (a) student A.
 - (b) student B.
 - (c) student C.
 - (d) all three student.

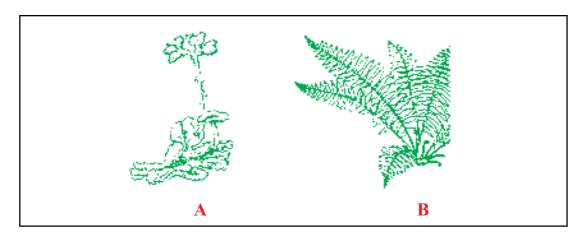


23. Four slides were observed under the microscope for spot test as shown below. The correct identification of the four spots is



(a)	A: onion peel	B: cheek cell	C: sclerenchyma	D: parenchyma
(b)	A: cheek cells	B: parenchyma	C: sclerenchyma	D: onion peel
(c)	A: parenchyma	B: cheek cells	C: onion peel	D: sclerenchyma
(d)	A: sclerenchyma	B: onion peel	C: parenchyma	D : cheek cells

24. Figures of two plants are given below. Observe them carefully and select the option which correctly gives their identification and the names of groups to which they belong.



- (a) A: moss; Bryophyta
 - B: Fern; Tracheophyta
- (b) A: pine; Gymnospermae
 - B: leafy plant; Angiospermae
- (c) Both A and B are liverworts and belong to Tracheophyta
- (d) Both A and B are ferns and belong to Bryophyta.

- 25. A student was given three samples containg ethanoic acid, sodium bicarbonate solution and water in test tubes I, II and III, respectively. On dipping a pH paper in them, he observed that the colour turned orange in I, blue in II and green in III. If arranged in inreasing order of their pH, the sequence of these bottles would be
 - (a) I, III, II.
 - (b) I, II, III.
 - (c) III, I, II.
 - (d) II, III, I.
- 26. On heating ferrous sulphate crystals, one would get
 - (a) Sweet smell
 - (b) lotten egg smell
 - (c) irritating, choking smell
 - (d) none of the above
- 27. In an experiment to trace the path of a ray of light passing through a rectangular glass slab, four students tabulated their observation as given below.

S.No.	Angle of incidence degree	Angle of refraction degree	Angle of emergence degree
1	30	18	32
2	45	28	43
3	60	35	60

A

S.No.	Angle of incidence degree	Angle of refraction degree	Angle of emergence degree
1	30	15	38
2	45	20	53
3	60	28	67

S.No. Angle of Angle of Angle of incidence refraction emergence degree degree degree C 1 30 10 31 2 45 44 15 3 60 22 60

S.No. Angle of Angle of Angle of incidence emergence refraction degree degree degree D 1 30 28 28 2 45 40 40 3 60 56 56

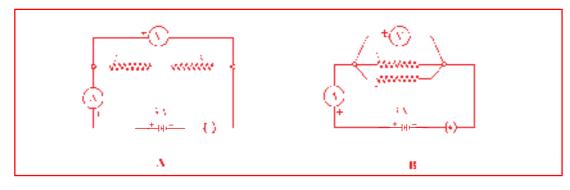
The student most likely to have done the experiment properly is

(a) A.

В

- (b) B.
- (c) C.
- (d) D.

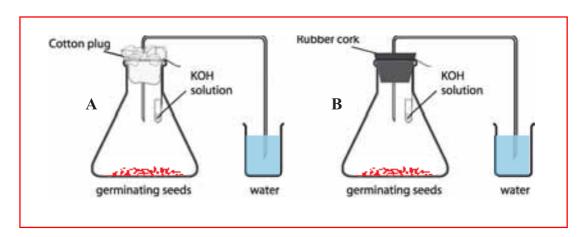
28. For the circuits A and B shown below, the voltmeter readings would be



- (a) 0.6 in circuit A and 2.5 V in circuit B.
- (b) 0 V in both circuits.
- (c) 5V in boh circuits.
- (d) 0 V in circuit A and 3 V in circuit B.
- 29. While performing an experiment with raisins, a student recorded the following data.

 $\begin{array}{ll} \text{Mass of water taken in the beaker} & = 50 \text{ g} \\ \text{Mass of raisins before soaking} & = 20 \text{ g} \\ \text{Mass of raisins after soaking} & = 30 \text{ g} \\ \text{Mass of water in beaker left after experiment} & = 40 \text{ g} \\ \text{The \% of water absorbed by the raisins is} & = 60 \text{ g} \\ \text{Mass of water in beaker left after experiment} & = 40 \text{ g} \\ \text{The \% of water absorbed by the raisins is} & = 60 \text{ g} \\ \text{Mass of water in beaker left after experiment} & = 60 \text{ g} \\ \text{Mass of water in beaker left after experiment} & = 60 \text{ g} \\ \text{Mass of water in beaker left after experiment} & = 60 \text{ g} \\ \text{Mass of water in beaker left after experiment} & = 60 \text{ g} \\ \text{Mass of water absorbed by the raisins is} & = 60 \text{ g} \\ \text{Mass of water absorbed by the raisins is} & = 60 \text{ g} \\ \text{Mass of water absorbed by the raisins is} & = 60 \text{ g} \\ \text{Mass of water absorbed by the raisins is} & = 60 \text{ g} \\ \text{Mass of water absorbed by the raisins is} & = 60 \text{ g} \\ \text{Mass of water absorbed by the raisins is} & = 60 \text{ g} \\ \text{Mass of water absorbed by the raisins is} & = 60 \text{ g} \\ \text{Mass of water absorbed by the raisins is} & = 60 \text{ g} \\ \text{Mass of water absorbed by the raisins is} & = 60 \text{ g} \\ \text{Mass of water absorbed by the raisins is} & = 60 \text{ g} \\ \text{Mass of water absorbed by the raisins is} & = 60 \text{ g} \\ \text{Mass of water absorbed by the raisins is} & = 60 \text{ g} \\ \text{Mass of water absorbed by the raisins is} & = 60 \text{ g} \\ \text{Mass of water absorbed by the raisins after ab$

- (a) 10%
- (b) 20 %
- (C)45%
- (d) 50 %
- 30. Using the same number of given germinating gram seeds, two students A and B set up the experiment separately. Student A used a cotton plug to hold the bent tube in the mouth of the flask. Student B used a rubber cork.



After 4 hours they noticed that

- (a) water level increased in the bent tube only of A.
- (b) water level increased in the bent tube only of B.
- (c) the cotton plug was wet.
- (d) the water in the beaker of tube turned milky.

Key for Sample Paper III

Q. No.	Correct Choice	Explanation/Remarks
1.	(a)	The larger is the size of particles of dispersed phase, lesser will be the stability of the inixture.
2.	(c)	To avoid side reaction and breakage at high temperature.
3.	(c)	Ammonium chloride sublimes and gets deposited on the cooler parts of the funnel and common salt does not sublime and remains unaffected leading to separation of the two compuounds.
4.	(d)	It is a displacement reaction in which whole of nail is in contact with copper sulphate solution, due to which reaction takes place anywhere on the nail.
5.	(d)	It is wrong to weigh a wet copper piece in air.
6.	(d)	The liquid cools faster first and slowly later on when its temprature gets close to the surrounding temprature.
7.	(a)	We would prefer using a pointed source and a pointed detector. The wooden tubes are needed for guiding the path of sound while the polished metal sheet acts as a good reflector of sound.
8.	(d)	The cotton string is a better carrier of the pulse while the stop clock enables us to measure time more accurately.
9.	(b)	Metanil yellow (aq) + HCl = Pink
10.	(b)	Mouth to be disinfected first and glycerine to be added last.
11.	(d)	Post anal tail is a chordate characteristic.
12.	(d)	Zinc reacts with sodium hydroxide on heating to produce hydrogen gas rapidly.
13.	(a)	Copper does not react, but zinc reacts with ferous sulphate solution. Thus zinc is most reactive, followed by iron and copper.
14.	(b)	The gases should not be smelt directly or kept too close to nose.
15.	(a)	The screen is just a device to observe the (real) image formed by a convex lens. The image of a distant object continues to get formed at focus of the convex lens even when no screen is being used to show its formation.
16.	(b)	The voltmeter has to be put in parallel with the resistances being measured and not across the ammeter.

- 17. (a) We must keep the ciruit closed for a relatively shorter time and open for a relatively longer time.

 18. (b) Nucleus divides first and then the cytoplasm, when Amoeba undergoes fission.

 19. (d) KOH absorbs CO₂ released by the seeds.
- 20. (b) Sunlight is not available to the covered portion, Hence no starch remains white after starch test.
- 21. (c) This arrangement, the set up is a closed arrangement with proper position of the glass tube fitted in a cork with two holes; the thermometer is inserted through the other hole.
- 22. (d) The overflow can must always be filled up to its spout before using it to measure the volume of water displaced by the immersed solid.
- 23. (d) correct identification of the four tissues
- 24. (a) correct identification and classification
- 25. (a) Ethonoic acid has the lowest pH and NaHCO₃ has the highest pH whereas the pH of water is in between the two.
- 26. (c) It is decompostion reaction is which SO₂ and SO₃ are liberated which are pungent smelling gases.
- We must not only have the angle of emergence (nearly) equal to the angle of incidence but also have an idea of the magnitude of the angle of refraction (for a glass slab) for the three most often used values (30°, 45°, 60°) of the angle of incidence.
- 28. (d) Only circuit B, with a dot within the symbol of the plug key, is a closed circuit in which current is flowing and will show non-zero voltage. The voltmeter reading, for the set ups shown, would be (nearly) equal to the voltage of the battery.
- 29. (d) Calculation using the formula.
- 30. (b) The conical flask should be closed with rubber cork to make it air tight. Otherwise partial vacuum will not be created.

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